

FENIKSOVA, R.V.; SEGAL, R.B.; RODZEVICH, V.I.; SHILOVA, A.A.

Aspergillus molds as producers of amylase. II. Amylase activity of Aspergillus oryzae as related to culture conditions on natural solid media. Mikrobiologiya 22, 145-50 '53. (CA 47 no.22:12523 '53) (MLRA 6:3)

1. All-Union Research Inst., Alcohol Ind., Moscow.

Feniksova, R. V.

Aspergillus as amylase producer. III. Influence of growth conditions on formation of conidia by *Aspergillus oryzae*. R. V. Feniksova and R. B. Segal (All-Union Sci. Research Inst. Microbiol., Moscow). *Microbiologiya* 22, 543-57 (1963); cf. *C.A.B.* 47, 12523d. When grown in layers (2.5-4.5 cm. thick) of wheat bran, *A. oryzae* does not form conidia throughout the layer unless porosity is increased by about 30% of inert matter, e.g., oat husks. The optimum atm. humidity for conidial growth is 90-95% and maintaining this humidity is a leading factor in forming conidia. Bimultation of CO<sub>2</sub> from the atm. is favorable, but not a decisive factor. Julian F. Smith

Feniksova, R. V.

The effect of the nutrient medium and of culturing procedures on the synthesis of amylolytic enzymes by *Aspergillus oryzae*. R. V. Feniksova, F. M. Kuzburskaya, and A. A. Sillova. *Trudy vuzovsk. Nauch.-Issledovatel. Inst. Spis. Khim.* 1954, No. 2, 110-27; *Referat. Zhur. Khim., Biol. Khim.* 1955, No. 12723. — A comparative study was made of 3 methods of culturing *A. oryzae*: (1) deep submerged, (2) surface, on solid medium, and (3) pellicular, on liquid medium. Method 3 proved most suitable for the purpose of obtaining higher amylolytic enzyme concn. Methods 1 and 2 yielded amylase of av. normal activity. The right concns and proper ratios of the constituents of the medium are requisites for the production of potent amylase. It is recommended that the concn. of N in the Czapek medium be increased 3-6 times and the concn. of the carbohydrates 2-3 times the original. Starch is the best source of carbohydrate and inorg.  $\text{NH}_4$  salts best source of N with the medium at a neutral pH. When synthetic medium was used in conjunction with starch and  $(\text{NH}_4)_2\text{SO}_4$ , method 1 proved most suited for the production of av. amylase.  $\text{NaNO}_3$  and  $\text{KNO}_3$  are of equal value as sources for N, but the amylase produced is of lower concn. With  $\text{NH}_4$  salts of org. acids high concn. of amylase was obtained with  $\text{H}_2\text{CO}_3$  and method 1. B. S. Levine.

FENIKSOVA, R.V.

New amylolytic ferments in molds. Spirt.prom.21 no.2:45 '55.  
(MIRA 8:10)

(Molds (Botany) (Fermentation))

Name: FENIKSOVA, Raisa Vasil'yevna

Dissertation: Amylolytic ferments of fungi of the  
genus Aspergillus

Degree: Doc Biol Sci

Affiliation: All-Union Sci Res Inst of Alcohol  
Industry

Defense Date, Place: 6 Jun 56, Council of Inst of Micro-  
biology, Acad Sci USSR

Certification Date: 16 Mar 57

Source: FMVO 13/57



FENIKSOVA, R. V.

3

7/16

1 Fermentation of farinaceous mashes by a thermophilic strain of yeast. R. V. Feniksova, L. G. Logikova and A. A. Shilova (*Mikrobiologiya*, 1958, 25, 310-317). - A strain of *Saccharomyces cerevisiae* XII adapted to growth at 37-38° was as effective in fermenting rye mash saccharified with malt at 38° as was the parent strain at 30°. Mashes saccharified with *aspergillus* or *rhizopus amylasea* were somewhat less efficiently fermented at 38°. R. Tauson.

FENIKSOVA, R. V.

(Acad. Sci. USSR)

"The Physiology of Nutrition of Aspergillus Oryzae in relation to the  
Formation of Active Amylase,"

paper submitted for presentation at the Intl. Symposium on Enzyme Chemistry,  
16-23 Oct. 1957, Tokyo, Japan

B-3,095, 529

**DECLASSIFIED**

PROCEEDINGS OF THE INTERNATIONAL SYMPOSIUM ON INZYME CHEMISTRY, Tokyo & Kyoto, 1957  
AUTHOR: Organizing Committee, International Symposium on Enzyme Chemistry  
Naruse, 1958.

M. V. FORTIN

supplied as ammonium phosphate, and with 40 mg total sugars as glucose (Table IV).

**Addendum**  
Purification of bacterial amylose and procrausine from the culture filtrate of *B. subtilis* was as follows: the amylose is specifically precipitated from the filtrate by ammonium sulfate from which it is washed by repeated centrifugation. The ammonium sulfate has been removed by washing with 10% per cent by weight aqueous sodium carbonate solution. The amylose is eluted from the sorbex by adding (15). The procrausine is eluted from the sorbex by adding with two volumes of water (40-50%). It is in this context, and not in the case of amylose, that the amylose is washed out by the filtrate after adsorption of amylose. A solution of the ammonium sulfate, washed, and then poured through a column of ion exchange resin (Rohm and Haas Co., 1941, 2, 10). The procrausine is eluted from the resin with a suitable eluting solvent. The amylose and procrausine thus obtained are readily crystallized.

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- Hayashi, T., and Matsushima, H. (1986). *Synopsis on Eclosion Chem. Japan* 59, 197 (1986).

# The Physiology of Nutrition of *Aspergillus oryzae* in Relation to the Formation of Active Amylase by this Fungus

**S. V. PAVLOV**

*Institute of Atomic Industry, Moscow 115250*

In most investigations of the physiology of *Aspergillus* under nutrition, determinations of the mycelial weight and of changes in the composition of the nutrient medium associated with growth were used as the chief criterion in assessing the suitability of some nutrient or other. The enzymic activity of the cultures was not investigated, and

It is only recently that papers have appeared in which nutritional requirements of certain microorganisms and the conditions of their cultivation have been studied as a function of the nature of the substrate. The growth of an organism on organic acids, antibiotics or enzymes. It is well known that the formation of enzymes by



FENIKSOVA, R. V.

Institute of Fermentation, Moscow.

"Mould Cultures as a Source of Enzyme Preparations,"

paper presented at Seventh International Congress of Microbiology, Stockholm,  
Sweden, 4 - 9 Aug '58.

IVANOV, I.D.; FENIKSOVA, R.V.

Polarographic determination of proteinase and amylase in *Bacillus subtilis*. *Biokhimiia* 24 no.2:222-224 Mr-Apr '59. (MIRA 12:7)

1. Institute of Biochemistry, Academy of Sciences of the U.S.S.R., Moscow.

(*BACILLUS SUBTILIS*, metab.  
amylase & protease, polarography (Rus))

(*AMYLASE*,  
in *Bacillus subtilis*, polarography (Rus))

(*PROTEASES*,  
same)

FENIKSOVA, R.V.

Alcohol, and the liqueur and vodka industry of the Chinese People's  
Republic. Spirt. prom. 25.no.7:19-22 '59. (MIRA 13:2)  
(China--Liquor industry)

FENIKSOVA, R.V.; DVADTSATOVA, Ye.A.

Amylolytic activity of *Aspergillus oryzae* grown on synthetic media. Trudy Inst. mikrobiol. no. 6:144-149 '59. (MIRA 13:10)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut spirtovoy promyshlennosti, Moskva.  
(*ASPERGILLUS ORYZAE*) (ENZYMES)

FENIKSOVA, R.V.; TIKHOMIROVA, A.S.

A medium for amylase and proteinase accumulation in submerged cultures of *Bacillus subtilis*. Mikrobiologiya 29 no.6:894-898 N-D '60. (MIRA 14:1)

1. Institut biokhimi imeni A.N. Bakha AN SSSR.  
(*BACILLUS SUBTILIS*) (PROTEINASE)  
(AMYLASE) (SOYBEAN)



FENIKSOVA, R. V. (USSR)

"Increase of Amylase Formation in a Stab Culture of *Aspergillus oryzae*."

Report presented at the 5th International Biochemistry Congress,  
Moscow, 10-16 Aug 1961

FENIKSOVA, R. V.; MOLODOVA, G. A.

Production of purified enzymatic preparations from surface mold  
fungi cultures. Mikrobiologiya 30 no.3:534-539 My-Je '61.  
(MIRA 15:7)

1. Institut biokhimii imeni A. N. Bakha AN SSSR.

(MOLDS(BOTANY)) (ENZYMES)

KONOVALOV, Sergey Aleksandrovich; LOGINOVA, L.G., doktor biol. nauk,  
retsenzent; ~~FENIKSOVA, H.V.~~, doktor biol. nauk, retsenzent;  
KOVALEVSKAYA, A.I., red.; KISINA, Ye.I., tekhn. red.

[Biochemistry of yeast] Biokhimiia drozhzhei. Moskva, Pishche-  
promizdat, 1962. 268 p. (MIRA 15:11)  
(Yeast) (Biochemistry)

BABAKINA, Vera Grigor'yevna; CHERNOV, N.V., doktor tekhn. nauk,  
prof., retsenzent; FENIKSOVA, R.V., doktor biol. nauk,  
retsenzent; PLEMYANNIKOV, M.N., red.; KNAKWIN, M.N., tekhn.  
red.; VINOGRADOVA, G.A., tekhn. red.

[Use of ferments in the manufacture of leather]Primenenie fer-  
mentov v proizvodstve kozhi. Moskva, Rostekhzdat, 1962. 239 p.  
(MIRA 15:12)

(Leather research)

(Fermentation)

FENIKSOVA, R.V.; GENDINA, S.B.

Production of more active varieties of mold fungi forming  
amylolytic enzymes. Trudy TSNIISP no. 13:18-25 '62.  
(MIRA 17:5)



FENIKSOVA, R.V.; SHILOVA, A.A.

Amylolytic enzymes of the *Aspergillus awamori* mold fungi. *Ferm.*  
i *spirt. prom.* 30 no.5:6-10 '64. (MIRA 17:10)

1. AN SSSR (for Feniksova). 2. Vsesoyuznyy nauchno-issledovatel'skiy  
institut fermentnoy i spirtovoy promyshlennosti (for Shilova).

FENIKSOVA, R.V.; RODIONOVA, N.A.; TIUNOVA, N.A.; ULEZLO, I.V.; SAFONOV, V.I.

Study of cellulolytic enzymes of *Myrothecium verrucaria*. Dokl. AN  
SSSR 162 no.3:702-704 My '65. (MIRA 18:5)

1. Institut biokhimii im. A.N.Bakha AN SSSR. Submitted August 17, 1964.

FENIKSOVA, V. V. Cand Geolog-Mineralog Sci

Dissertation: "Stratigraphy of Red-Colored Deposits of the Area Along  
The Volga River in Gor'kiy Oblast." Moscow Order of Lenin State U  
imeni M. V. Lomonosov 19 Jun 47

SO: Vechernyaya Moskva, Jun 1947 (Proj #17836)

FENIKSOVA, R.V.; PETROVA, I.S.; VINTSYUNAYTE, M.M.; BABKINA, V.G.;  
~~NESTEROVA, G.A.~~

Use of proteolytic enzymes of *Actinomyces fradiae* for removal of  
wool from raw hides. Prikl. biokhim. i mikrobiol. 1 no.3:257-262  
My-Je '65. (MIRA 18:7)

1. Institut biokhimii AN SSSR imeni Bakha.

FENIKSOVA, R.V.; SHILOVA, A.A.

Formation of enzymes in the surface culture of *Aspergillus*  
awamori. *Ferm. i spirt. prom.* 31 no.2:4-5 '65.

(MIRA 18:6)

1. AN SSSR (for Feniksova). 2. Vsesoyuznyy nauchno-issledovatel'skiy  
institut fermentnoy i spirtovoy promyshlennosti (for Shilova).



FENIKSOVA, R.V.; PETROVA, I.S.

Proteolytic enzymes in *Actinomyces fradiae*. Prikl. biokhim. i  
mikrobiol. 1 no.2:175-180 Mr-Apr '65.

(MIRA 18:11)

1. Institut biokhimii imeni A.N.Bakha AN SSSR, Moskva.

FENIKSOVA, R.V.; ULEZLO, I.V.

Study of the biosynthesis of cellulase by *Myrothecium verrucaria*.

Prkl. biokhim. i mikrobiol. i no. 4:406-413 J1-Ag '65.

(MIRA 18:11)

I. Institut biokhimii imeni A.N.Bakha AN SSSR.

**FEKISOVA, V.V.**

Stratigraphic role of the "dark blue argillaceous" horizon in  
the Cenozoic of the middle Ob' Valley. *Biul.MOIP. Otd.geol.*  
31 no.4:45-54 J1-Ag '56. (MIRA 9:12)

(Ob' Valley--Geology, Stratigraphic)

FENIKSOVA, V.V.; DUBROVO, I.A.

Terraces of the Yenisey Valley at the mouth of the Kan River and  
their geological age. Vest.Mosk.un.Ser.biol., pochv., geol., geog.  
14 no.1:121-130 '59. (MIRA 12:9)

1. Moskovskiy gosudarstvennyy universitet, Kafedra istoricheskoy  
geologii.  
(Yenisey Valley--Geology, Stratigraphic)

BROD, I.O., prof., doktor geol.-miner. nauk; VARSANOF'YEVA, V.A.,  
 prof., doktor geol.-miner. nauk; VELIKOVSKAYA, Ye.M., prof.,  
 doktor geol.-miner. nauk; GORDEYEV, D.I., prof., doktor  
 geol.-miner. nauk; DOBROV, S.A., doktor geol.-miner. nauk  
 [deceased]; KOF, M.I., kand.tekhn.nauk, [deceased]; KUZMICHEVA,  
 Ye.I., mladshiy nauchnyy sotr.; KUZNETSOV, Ye.A., prof., doktor  
 geol.-miner. nauk; LEONOV, G.P., prof., doktor geol.-miner. nauk;  
 MENNER, V.V., dotsent, doktor geol.-miner. nauk; NAZARENKO, I.I.,  
 kand. sel'khoz.nauk; POBEDIMSKAYA, Ye.A., assistant; POPOV, S.P.,  
 prof., doktor geol.-miner. nauk; SMIRNOV, V.I.; SMIRNOV, N.N.,  
 prof., doktor geol.-miner. nauk; SMOL'YANINOV, N.A., prof.,  
 doktor geol.-miner. nauk [deceased]; FENIKSOVA, V.V., dotsent,  
 kand.geol.-miner. nauk; SHAFRANOVSKIY, I.I., prof., doktor geol.-  
 miner. nauk; Primalni uchastiye: BARSANOV, G.P., prof.,  
 doktor geol.-miner. nauk; BOKIY, G.B.; GORSHKOV, G.P., prof.,  
 doktor geol.-miner. nauk; KUDRYAVTSEV, V.A., prof., doktor  
 geogr. nauk; MARKOV, P.N., dotsent, kand.geol.-miner. nauk;  
 MOROZOV, S.S., prof., doktor geol.-miner. nauk; ORLOV, Yu.A.,  
 akademik; SERGEYEV, Ye.M., prof., doktor geol.-miner. nauk;  
 TVALCHRELIDZE, A.A.; GEORGIYEVA, G.I., tekhn. red.

(Continued on next card)



BROD, I.O.--- (continued) Card 2.

[History of geology at Moscow University] Istoriiia geologicheskikh nauk v Moskovskom universitete. Pod red. D.I.Gordeva. Moskva, Izd-vo Mosk. univ., 1962. 351 p. (MIRA 15:7)

1. Moscow. Universitet. Geologicheskii fakul'tet. 2. Chlen-korrespondent Akademii nauk SSSR (for Smirnov). 3. Chlen-korrespondent Sibirskogo otdeleniya Akademii nauk SSSR (for Boki ). 4. Deystvitel'nyy chlen Akademii nauk Gruzinskoy SSR (for Tvalchrelidze).

(Moscow University) (Geology--Study and teaching)

FENIKSOVA, V.V.

New finds of Quaternary mammal fauna in the Chulyms Valley.  
Geol. i geofiz. no.3:156-158 '65. (MIRA 13:6)

1. Moskovskiy gosudarstvennyy universitet.

FENIN, N. K.

189T49

USSR/Hydrology - Irrigation

Aug 51

"The Digging of Temporary Irrigation System  
Taking Into Consideration Counter-Filtration  
Measures," Asst Prof N. K. Fenin

"Gidrotekh i Meliorat" No 8, pp 23-39

Fenin presents technological scheme for sub-  
ject adapted to natural conditions of the  
Volga region. He compiles tables of amt of  
water saved by counter-filtration measures  
and outlines soil conditions under which  
counter-filtration becomes indispensable.

189T49

FENIN, N.K., kandidat tekhnicheskikh nauk.

Hydromechanical method of preventing seepage from canals by artificial  
silting. Gidr. i mel. 5 no.6:36-48 Je '53. (MLRA 6:7)  
(Canals)

*F. C. N. K.*  
AUTHOR: Penin, N.K., Engineer

99-58-3-9/12

TITLE: The Danube - Tissa - Danube Canal, Yugoslavia  
(Kanal Dunay - Tissa - Dunay, Yugoslaviya)

PERIODICAL: Gidrotekhnika i Melioratsiya, 1958, # 3, pp 42-48 (USSR)

ABSTRACT: This article describes the proposed large-scale building of canals in Yugoslavia. The system of canals will cover the territory of Bachka and Banat, which is the most fertile part of Yugoslavia. Already, 10 years of surveying and preparatory work have gone into this project. The diversity of geologic structures has complicated the work, although some parts of this canal-system already exist. At present, there are 1,234 km of dikes protecting over 500,000 hectares of land against floods, as well as a system of shallow canals for the draining of swamps. The total meliorated area amounts to 761,000 hectares, 80% of which is arable land. The irrigated land occupies only 5,000 hectares. There are also 266 km of navigable canals, which were constructed during the last 2 centuries. The new plan, which will take 10 years to be realized, has 3 aims: 1) Construction of an additional 62,5 km of flood protecting banks; 2) Melioration of 761,000 of hectares; 3) Construction of an additional 400 km of water ways. Two hydro-electric

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The Danube - Tissa - Danube Canal, Yugoslavia

99-58-3-9/12

power stations will be built - one on the Tissa river with a capacity of 4,400 kw; the second on the Banat canal with a capacity of 740 kw. It is calculated, that as a result of the realization of this project, the yields of all agricultural crops will be increased 3,6 times.

There are 4 figures and 1 map.

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Card 2/2

FENIN, N.K., inzh.

Effect of artesian waters on the stability of canal profiles.  
Gidr. 1 mel. 13 no.4:52-57 Ap '61. (MIRA 14'4)

1. Moskovskaya ordena Lenina sel'skokhozyaystvennaya akademiya  
im. K.A.Timiryazeva.  
(Yugoslavia—Canals) (Water, Underground)

FENIN, Nikolay Konstantinovich; YASINETSKIY, Vyacheslav Grigor'yevich;  
Prinimal uchastiye MER, I.I.; BERKOV, A.M., kand. tekhn.nauk,  
retsenzent; DROBYSHEV, G.I., kand. tekhn. nauk, retsenzent;  
MINKIN, V.I., kand. tekhn. nauk, retsenzent; SHIMANOVICH, V.S.,  
inzh., retsenzent; YELIZAVETSKAYA, G.V., red.; MAKHOVA, N.N.,  
tekhn. red.

[Organization and technology of irrigation and drainage  
construction work] Organizatsiia i tekhnologiya gidromelio-  
rativnykh rabot. Moskva, Sel'khozizdat, 1963. 478 p.

(MIRA 17:1)

1. Kafedra stroitel'nogo proizvodstva i mekhanizatsii Novo-  
cherkasskogo inzhenerno-meliorativnogo instituta (for Berkov,  
Drobyshev, Minkin). 2. Gosudarstvennyy Komitet Soveta Ministrov  
RSFSR po vodnomu khozyaystvu (for Shimanovich).



FENIN, N.K., inzh.; KATKEVICHYUS, L.A. [Katkevicius, L.], inzh.

Improvement of excessively wet soils by hydraulic filling.  
Gidr. i mel. 17 no.5:30-38 My '65. (MIRA 18:7)

LI, VAN SEN [Yi, Wan-son]; MIKHAYLOV, I.N.; FENIN, Yu.I.; SARANTSEVA,  
V.R., tekhn. red.

[Reaction  $T + T \rightarrow He^4 + 2n$ ] O reaktsii  $T + T \rightarrow He^4 + 2n$ .  
Dubna, Ob"edinennyi in-t iadernykh issl., 1962. 8 p.

(MIRA 15:4)

(Nuclear reactions)

POPOV, Yu.P.; FENIN, Yu.I.; SARANTSEVA, V.R., tekhn. red.

[Analysis of averaged cross sections of neutron capture]  
Analiz usrednennykh sechenii zakhvata neitronov. Dubna,  
Ob"edinennyi in-t iadernykh issl., 1962. 14 p.  
(MIRA 15:10)

(Neutrons--Capture)

44222  
S/056/62/043/006/005/067  
B163/B186

26.2245  
AUTHORS:

Popov, Yu. P., Fenin, Yu. I.

TITLE:

Analysis of average cross sections for neutron capture

PERIODICAL:

Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 43,  
no. 6(12), 1962, 2000 - 2007

TEXT: Experimental data on the energy dependence of average neutron capture cross sections are analyzed, these having been collected in the Fizicheskii institut im. P. N. Lebedeva (Physics Institute imeni P. N. Lebedev) using a lead slowing down time neutron spectrometer. The analysis is based on equation

$$\bar{\sigma}_v = \sum_{l,j} \bar{\sigma}_v^{l,j} = 2\pi^2 k^2 \sum_{l,j} \frac{2j+1}{2(2l+1)} \sum_i \frac{\Gamma_n^{l,j} \Gamma_\gamma^{l,j}}{D^2 \Gamma^{l,j}} \quad (1)$$

i. e. the Breit-Wigner cross section formula for an isolated resonance, averaged over  $l, J, j$ . In (1),  $\lambda$  is the neutron wave length,  $l$  the spin of the target nucleus and  $J$  that of the compound nucleus,  $j = l \pm \frac{1}{2}$  the total momentum of the neutron;  $\Gamma_n^{l,j}$ ,  $\Gamma_\gamma^{l,j}$ , and  $\Gamma^{l,j}$  respectively are the

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S/056/62/043/006/005/067  
B163/B186

Analysis of average ...

partial neutron width corresponding to a given value  $j$ , the radiation and the total widths of the resonance level;  $D^j$  is the average spacing between levels having equal  $J$ . The vinculum in equation (1) denotes averaging over width distributions from resonance to resonance where the Porter-Thomas distribution is used for neutron widths. Inelastic scattering is neglected. The number of unknown parameters in equation (1) is reduced under the following assumptions. The level density  $\rho = 1/D^j$  and  $\Gamma_n$  are considered to be independent of energy. For the dependence of  $\rho$  and  $\Gamma_n$  on  $J$  the description by the statistical model is used; this means that  $\Gamma_n$  may be considered practically independent of  $J$ .  $\langle \Gamma_n(1J) \rangle$  is defined by  $\Gamma_n^{1J} = \langle \Gamma_n(1J) \rangle \xi_J^{11}$  where  $\xi_J^{11}$  is equal to 2 if  $|J - I| \leq 1 + \frac{1}{2} \leq J + I$ , and equal to 1 if only one of the conditions  $|J - I| \leq 1 + \frac{1}{2} \leq J + I$  or  $|J - I| \leq 1 - \frac{1}{2} \leq J + I$  is fulfilled; otherwise  $\xi_J^{11} = 0$ .  $\langle \Gamma_n(1)/D \rangle$  is defined by averaging  $\langle \Gamma_n(1J) \rangle / D^j$  over  $J$ . With these definitions, the following equation results:  $\bar{\sigma}_v = 2\pi^2 k^2 \sum_{J,I} \frac{2J+1}{2(2I+1)} \frac{\xi_J^{11} \langle \Gamma_n(1)/D \rangle (\Gamma_v^1/D^j)}{\xi_J^{11} \langle \Gamma_n(1)/D \rangle + (\Gamma_v^1/D^j)} F \left( \frac{\Gamma_v^1/D^j}{2\xi_J^{11} \langle \Gamma_n(1)/D \rangle} \right) \quad (2)$

Card. 2/ 5

Analysis of average ...

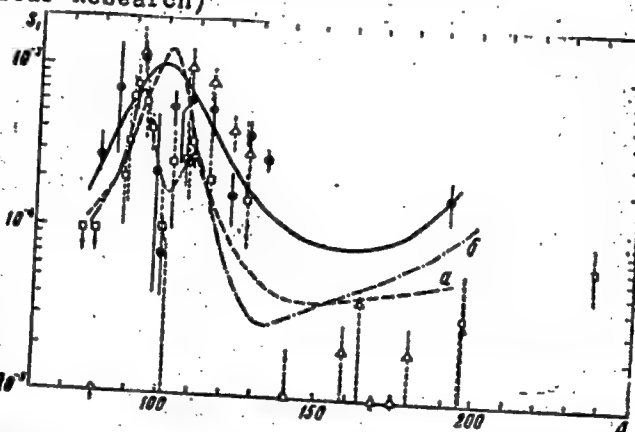
S/056/62/043/006/005/067  
B163/B186

the other parameters vary considerably. In Fig. 3 the p wave strength function is compared with results by other authors. There are 3 figures and 1 table.

ASSOCIATION: Ob'yedinennyy institut yadernykh issledovaniy (Joint Institute of Nuclear Research)

SUBMITTED: June 20, 1962

Fig. 3: Dependence of the strength function for p wave neutrons on the atomic weight of the target nucleus. ● - Results of the present work. □ - taken from Weston et al., Δ - from Gibbons et al. The solid curve is calculated from the sticking coefficients used by Nemirovskiy (potential with diffuse boundary, volume absorp-  
Card 4/5



Analysis of average ...

S/056/62/043/006/005/067  
B163/B186

tion). The dotted curves are taken from Krueger and Margolis (potential with diffuse boundary, surface absorption). Curve  $\sigma$  corresponds to twice the spin-orbital interaction as compared with  $\alpha$  and the solid curve. The signs with arrows show upper limits of  $S_1$ .

Card 5/5

ACCESSION NR: AP4043660

S/0056/64/047/002/0777/0778

AUTHORS: Fenin, Yu. I.; Shapiro, F. L.

TITLE: On the connection between the scattering length and the neutron radiative capture cross section

SOURCE: Zh. eksper. i teor. fiz., v. 47, no. 2, 1964, 777-778

TOPIC TAGS: scattering length, radiation width, capture cross section, elastic scattering, scattering amplitude, even even nucleus

ABSTRACT: It is shown that in the case of low-energy neutrons, when the main contribution to the interaction with nuclei comes from the partial wave for zero angular momentum, a simple relation exists between the cross sections for elastic scattering and for radiative capture of neutrons in the region between resonances, in the form

(1)

Card 1/2



ACCESSION NR: AP4043660

Here  $\sigma_J$  -- total cross section for radiative capture of neutrons in the channel with spin  $J$ ;  $g_J$  -- statistical weight for the  $J$  channel;  $\lambda$  -- neutron wavelength;  $\Gamma_\gamma$  -- radiation width;  $a_J$  -- scattering amplitude for the  $J$  channel. A proof of the formula is presented and it is concluded that this relation can be used to determine radiation widths and capture cross sections for measurements of the total cross sections of even-even nuclei. In some cases the use of this formula can yield valuable information for odd nuclei also. In particular, it is reported that it has been applied to obtain the radiation widths of  $Cl^{35}$  and  $Sc^{45}$ . Orig. art. has: 5 formulas.

ASSOCIATION: Ob'yedinennyy institut yadernykh issledovaniy (Joint Institute of Nuclear Research)

SUBMITTED: 15May64

SUB CODE: NP

NR REF SOV: 000

ENCL: 00

OTHER: 002

Card 2/2

FENINA, Ye.P.

Dynamics of the change in the content of  $\gamma$ -globulins in the process of immunization of pregnant women with staphylococcal sorbed anatoxin. Akush. i gin. 40 no.1:36-38 Ja-F '64.

(MIRA 17:8)

1. Biokhimicheskaya laboratoriya (zav. - dotsent V.A. Yur'yev)  
i 2-ye akusherskoye otdeleniye (zav. - prof. S.G. Khaskin)  
Instituta akusherstva i ginekologii (dir. - prof. M.A. Petrov-Maslakov) AMN SSSR, Leningrad.

S/627/60/002/000/025/027  
D299/D304

3.2410

AUTHORS: Penivash, E., Frenkel', A., Telbits, F., Pernegr, Ya.,  
Petrzhilka, V., Sedlak, Ya., and Vrana, I.

TITLE: Investigating high-energy electron-photon cascade in  
emulsions

SOURCE: International Conference on Cosmic Radiation. Moscow,  
1959. Trudy. v. 2. Shirokiye atmosferye livni i kas-  
kadnyye protsessy, 307-310

TEXT: The energy spectrum of the primary photon was determined;  
the energy spectrum of pairs formed at depths of up to 1.5 units  
was studied. The obtained spectra were compared with the distribu-  
tion based on Bethe-Heitler's theory, and with that based on Migdal's  
formulas (a further development of the Landau approximation). The  
energy  $E_0$  of the primary photon was determined by the Chudakov-Per-  
kins effect, by the longitudinal and lateral shower development,  
and also by Pinkau's method. The values for the primary energy,

Card 1/ 3

Investigating high-energy ...

S/627/66/002/000/025/027  
D299/D304

obtained by shower development in the approximations A and B, were underrated. A more accurate energy estimate is obtained by means of the curves of A. A. Varfolomeyev and I. A. Svetloolobov (Ref. 11: ZhETF, 36, 1771, 1959). The data of Ref. 11 yielded a higher value for the primary energy. In the following, a primary energy of  $2 \cdot 10^{12}$  ev. is assumed. The energy of electron pairs was determined either by E. Lohrmann's method (Ref. 15: Nuovo Cim., 2, 1029, 1955) or by measuring multiple scattering. In some cases both methods were used. The results are shown in a table and in 2 figures which also exhibit (for comparison) two theoretical curves corresponding to Bethe-Heitler's and Migdal's formulas, respectively. The authors conclude that by studying only one or a few cascades, no definite decision can be made as to the validity of either Bethe-Heitler's or Landau-Migdal's theory. In this light, the present investigation should be considered as a contribution to the general statistics of cascades, investigations of a larger number of shower cascades being required before reaching a definite conclusion. The authors express their thanks to Professors Yanoshi, Farkas and Danysh. There

Card 2/3

Investigating high-energy ...

S/627/60/002/000/025/027  
D299/D304

are 2 figures, 2 tables and 18 references: 12 Soviet-bloc and 6 non-Soviet-bloc. The references to the English-language publications read as follows: D. H. Perkins, Phil. Mag., 46, 1146, 1955; K. Pin-  
kau. Phil. Mag., 2, 1389, 1957; J. C. Butcher, B. A. Chartres and  
H. Messel. Nuc. Phys., 6, 271, 1958; J. Nishimura and K. Kamata,  
Prog. Theor. Phys., 7, 185, 1952.

ASSOCIATION: Tsentral'nyy issledovatel'skiy institut fiziki, otde-  
leniye kosmicheskikh luchey (Central Research Insti-  
tute of Physics, Cosmic Ray Section, Budapest); Fi-  
zicheskiy institut Akademii nauk (Physics Institute  
of the Academy of Sciences, Prague)

Card 3/3

FENJE, Pavle, dr.

Outbreak of typhoid fever in a college at Sremska Mitrovica. Glasn.  
hig. inst., Beogr. 3 no.1-2:42-45 Jan-June 54.  
(TYPHOID FEVER, epidemiol.  
in Yugos.)

L 04303-67 EWI(1)/T-2 FDN/WW

ACC NR: AP6005388

(N)

SOURCE CODE: UR/0413/66/000/001/0139/0139

AUTHORS: Reka, Ya. D.; Khudyakov, Ye. D.; Chernobay, I. F.; Fenkel'shteyn, L. A.; Kultygin, N. S.; Lavrenyuk, N. A.

ORG: none

54  
B

TITLE: A pneumatic drive direct-action pump pressure booster. Class 59, No. 177772  
/announced by Donets State Design-Construction and Experimental Institute of the  
Complex Mechanization of Mines (Donetskiy gosudarstvennyy proyektno-konstruktorskiy  
i eksperimental'nyy institut kompleksnoy mekhanizatsii shakht)7

SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 1, 1966, 139

TOPIC TAGS: water pump, high pressure pump, high pressure pneumatic device,  
hydraulic pressure amplifier

ABSTRACT: This Author Certificate presents a pneumatic drive direct-action double  
acting pump pressure booster. The device includes a pneumatic cylinder with a  
piston, two operating cylinders with pistons rigidly connected with the piston of  
the pneumatic cylinder, and a distributing valve which is repositioned with the aid  
of checking devices when the piston approaches the extreme piston (see Fig. 1).  
The design increases the lifetime of the pump. The piston of the pneumatic cylinder  
is equipped at its ends with blades for rotating the piston to a specified angle

Card 1/2

UDC: 621.651.002.54

L 04303-67

ACC NR: AP6005388

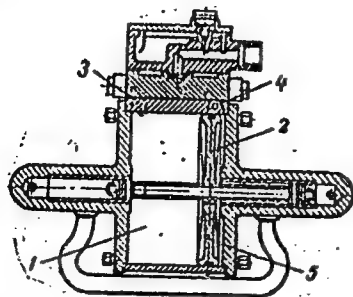


Fig. 1. 1 - pneumatic cylinder; 2 - piston; 3 and 4 - checking devices; 5 - blades

with each stroke. Orig. art. has: 1 figure.

SUB CODE: 13/

SUBM DATE: 18Jan64

Card 2/2 *gl*



BANG, T. [Bang, Th.]; FENKHEL, V. [Fenchel, W.]

Solving a covering problem of convex bodies. Mat. pros. no.1:214-218  
'57.

(Geometry)

(MIRA 11:7)

GRISHKO, N.P.; FENKINA, R.A.

Outbreak of alimentary toxicoinfection caused by staphylococcus and  
Escherichia coli. Med.zhur.Uzb. no.1:59:60 Ja '59. (MIRA 13:2)

1. Iz klinicheskoy infektsionnoy bol'nitsy No.1 goroda Samarkanda  
(glavnyy vrach - N.N. Aronbayev).  
(SAMARKAND--FOOD POISONING) (STAPHYLOCOCCUS) (ESCHERICHIA COLI)

1373 A kinetic method for determining the  $\gamma$ -isomer of hexachlorocyclohexane in tests of concentrated preparations. Fen'kova, S. *Russ. Khim.*, 1955, (155), 20-30; *Eng. Chem. Khim.*, 1955, Abstr. No. 43,582. — For elimination of the non-reproducibility of results obtained in determining the  $\gamma$ -isomer by an earlier method (Chur, *Anal. Chem.*, 1945, 20, 241) it is suggested that quantities of ethanolic soln. of alkali free from  $\text{CO}_2$  and equivalent to 10 ml of  $N$  soln. are cooled in an ice bath and added to tubes containing cooled ethanolic soln. of weighed samples. Blank soln. are also prepared without the addition of alkali. After standing in the ice bath for 15 min., the alkali is added and stirred continuously for 3 to 10 min. and then periodically during 60 min. Chloride is determined by titration with  $\text{AgNO}_3$  and the percentage of  $\gamma$ -isomer is found from the formula:  $x = 25.3(a - b) - 14.8$ , where  $a$  is the vol. (ml) of 0.1  $N$   $\text{AgNO}_3$  soln. used up; the second sample,  $b$  is the same for the blank. The reproducibility of results is  $\pm 0.05$  ml of 0.1  $N$   $\text{AgNO}_3$ . The divergence from results obtained by the chromatographic method is  $< 1\%$  for technical products, 0.5 to 2.5% for conc. products and 0.1 to 0.15% for dusts.

R. LORR

SHOGAM, S.M.; FEN'KOVA, Ye.I.; GAR, K.A.; POSLAVSKIY, Yu.M.; GOLUBEVA, Z.Z.

Investigation of fillers and selection of appropriate machinery  
for the production of new organic powder insecticides. [Trudy]  
NIUIF no.164:3-5 '59. (MIRA 15:5)

(Insecticides)

SHOGAM, S.M.; FEN'KOVA, Ye.I.; EPSHTEYN, T.B.

Physicochemical methods for determining the  $\gamma$ -isomer of  
hexachlorocyclohexane in various preparations. [Trudy] NIUIF  
no.164:35-36 '59. (MIRA 15:5)

(Benzene hexachloride)

SLOAN, S.M., kand.khimicheskikh nauk; FEN'KOV, Ye.I.; YEPHOREV, I.A.;  
EPSTEIN, T.B.

Insecticide powders, dusts and granulated insecticides. Zhur.  
VKHO 5 no. 3:312-317 '60. (MIRA 14:2)  
(Insecticides)

SHOGAM, S.M.; ORLOV, V.I.; FEN'KOVA, Ye.I.

Mineral substances used as fillers for powdered insecticides.

Trudy IGEM no.95:113-119 '63.

(MIRA 16:12)

BRYKCYNSKA, M.; FENKOWSKI, A.; KLAMBOROWSKI, J.

Center of Scientific Information and Publications in the Institute  
of Computers of the Polish Academy of Sciences. Akt probl inf  
dok 7 no.3:52-58 My-Je '62.



FENNELONOVA, Z. V., ROZEN'YER, L. A., and PRYANISHNIKOVA, V. T.

FENNELONOVA, Z. V., ROZEN'YER, L. A., and PRYANISHNIKOVA, V. T. "The treatment of diphtheria with small doses of serum in combination with sulfadine", Trudy Kishinevsk. gos. med. in-ta, Vol. 1, 1949, p. 122-27.

SO: U-3261, 10 April 53 (Letopis - Zhurnal 'nykh Statey No. 11, 1949)

ACC NR: AP7003336

SOURCE CODE: UR/0076/66/040/612/3092/3094

AUTHOR: Gordiyenko, S. P.; Fesenko, V. V.; Fenchka, B. V.

ORG: Institute of Materials Science Problems, AN UkrSSR (Institut problem materialovedeniya AN UkrSSR)

TITLE: Vapor composition and heats of vaporization of cerium, samarium, gadolinium and terbium hexaborides

SOURCE: Zhurnal fizicheskoy khimii, v. 40, no. 12, 1966, 3092-3094

TOPIC TAGS: heat of vaporization, cerium compound, samarium compound, gadolinium compound, terbium compound, boride, heat of dissociation

ABSTRACT: In order to arrive at a definitive explanation of the nature of vaporization and vapor composition over rare earth hexaborides, the authors studied the vaporization of  $CeB_6$ ,  $SmB_6$ ,  $GdB_6$ , and  $TbB_6$  using the apparatus and techniques employed previously, but also using Langmuir's method to produce a molecular beam. In each case, the spectra of ions in the range of 10-200 a.m.u. and 1900-2300°K showed only atomic ions of the lanthanides of the original hexaborides. At higher temperatures (2200-2500°K),  $11B^+$  and  $10B^+$  ions were observed, the ratio of boron-to-metal atom concentrations being no higher than 4:1. Curves of ion current intensity versus the energy of ionizing electrons were plotted and found to be linear, and the appearance potentials coincided with the ionization potentials of the elements, indicating the absence

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UDC: 541.11

ACC NR: AP7003336

of dissociative origin of the ions recorded. The data show that rare earth hexaborides at 1900-2300°K dissociate in accordance with the reaction



The heats of this reaction,  $\Delta H_T^0$ , for the hexaborides studied were calculated from the dependence of  $\log (IT)$  on  $1/T$  by the least-squares method. The lowest heat of dissociation, that of  $\text{SmB}_6$ , is 103 kcal/mole, and that of  $\text{CeB}_6$ ,  $\text{GdB}_6$  and  $\text{TbB}_6$  is 124, 128 and 129 kcal/mole respectively. An attempt is made to correlate these values with the electronic structure of the rare earth elements. Orig. art. has: 1 figure, 1 table and 2 formulas.

SUB CODE: 07/ SUBM DATE: 14Jan66/ ORIG REF: 004/ OTH REF: 001

Card 2/2

L 1649-66 ENT(m)/ENT(t)/ENT(b) IJP(c) JD/JW/JG

ACCESSION NR: AF5021425

UR/0073/65/039/003/2049/2051  
541.1

AUTHOR: L'vova, A. S.; Feodos'yev, N. N.

TITLE: Enthalpy of formation of calcium, strontium, and barium metahafnates

SOURCE: Zhurnal fizicheskoy khimii, v. 39, no. 8, 1965, 2049-2051

TOPIC TAGS: enthalpy, calcium metahafnate, strontium metahafnate, barium metahafnate

ABSTRACT: The enthalpies of the reactions



were measured by using a bomb calorimeter and an initiator (carbon black). The reaction products were identified by x-ray analysis. The reaction of  $\text{BaCO}_3$  and  $\text{SrCO}_3$  with  $\text{HfO}_2$  was not associated with the thermal decomposition of the unreacted carbonate, but  $\text{CaCO}_3$  did decompose. Since the reactions took place at constant volume, in

Card 1/2

L 1649-66

ACCESSION NR: AP5021425

passing from  $\Delta U$  to  $\Delta H$ , a correction was introduced which was equal to  $RT$  per mole of  $\text{CO}_2$  evolved. The enthalpies obtained for the reactions  $\text{MeCO}_3 + \text{HfO}_2 \rightarrow \text{MeHfO}_3 + \text{CO}_2$ , where  $\text{Me} = \text{Ca}, \text{Sr}, \text{and Ba}$ , were used to calculate the standard enthalpies of formation of calcium, strontium, and barium metahafnates from the oxides and elements. The data of O. Kubaschewski and E. L. L. Evans (Metallurgische Thermochemie, VEB Verlag Technik, Berlin, 1959) were used in the calculations. Orig. art. has: 5 tables.

ASSOCIATION: Rostovskiy na-Donu gosudarstvennyy universitet (Rostov-on-Don-State University)

SUBMITTED: 13Jul64

ENCL: 00

SUB CODE: GC

NO REF SOV: 003

OTHER: 003

Card 2/2 *SP*

FENOGENOV, A.N.

New data on the mineralogy of the Dagtyarka copper-pyrite deposit.  
Nauch.dokl.vys.shkoly; geol.-geog.nauki no.1:155-156 '58.

(MIRA 12:2)

1. Moskovskiy universitet, geologicheskiy fakul'tet, Ural'skaya  
geologicheskaya ekspeditsiya.  
(Ural Mountains—Ore deposits)

FENOGENOV, A.N.

~~Epileucite~~ porphyries in the central Urals. Izv. ucheb. zav.; geol.  
i razv. 1 no.7:56-62 J1 '58. (MIRA 12:8)

1. Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova,  
Kafedra petrografii.  
(Ural Mountains--Epileucite)

FEINGENOV, A.N.

Genesis of the Degtyarka copper pyrite deposit in the central Urals.  
Izv.vys.ucheb.zav.; geol.i razv. 2 no.8:98-115 Ag '59.  
(MIRA 13:4)

1. Moskovskiy gosudarstvennyy universitet.  
(Degtyarka region (Ural mountains)--Chalcopyrite)



FENOGENOV, A.N.

Relationship between vein granites and pyritic ore formation. Biul.  
» MOIP.Otd.geol. 35 no.2:156-157 M-Ap '60. (MIRA 14:4)  
(Ural Mountains--Pyrites) (Ural Mountains--Granite)

FENOGENOV, A.N.

Characteristics of the mineral composition of ores in the Mank deposit  
of the Central Urals. Vest.Mosk.un.Ser.4: Geol. 17 no.1:50-61  
Ja-F '62. (MIRA 15:2)

1. Kafedra petrografii Moskovskogo universiteta.  
(Ural Mountains-- Ore deposits)

FENOGENOV, A.N.

Zonal pyrite from the ores of the Yel'nichnyi section and Zyuzel'skoye deposit in the Central Urals. Izv.vys.ucheb.zav. geol. i razv. 6 no.5:78-82 My '63. (MIRA 18:4)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.

FENOGENOV, A.N.

Inclusion of chalcopyrite, sphalerite and other minerals in  
the pyrite of certain sulfide deposits in the Central Urals.  
Izv. vys. ucheb. zav.; geol. i razv. 6 no.12:62-65 D '63  
(MIRA 18:2)

1. Moskovskiy gosudarstvennyy universitet.

MOSKALEV, V.S.; FENOGEIKOV, I.V.; VASIL'YEV, B.D.

Frost action under foundation during the construction of buildings. Osn., fund.i mekh.grun. no.6:20-22 '59.

(MIRA 13:4)

(Foundations) (Frozen ground)

MOSKALEV, V.S., inzh.; FENOGENOV, I.V., inzh.

Prevent soil freezing in foundation beds of buildings. Rech.  
transp. 18 no.6:43-45 Je '59. (MIRA 12:9)  
(Soil freezing) (Foundations)

ACC NR: AP6013520

UR/0120/66/000/002/0169/0173

AUTHOR: Goryunov, N.N.; Ovechkin, Yu.A.; Tolkacheva, Ya.A. Feoktistov, Yu.F.

ORG: None

TITLE: Observation of heat fields in semiconductor devices

SOURCE: Priboiy i tekhnika eksperimenta, no.2, 1966, 169-173

TOPIC TAGS: transistor, transistor temperature, temperature sensing film, semiconductor device, heat sensing fluorescent film, fluorescent compound / K-9 fluorescent compound / FKP-03K fluorescent compound / FK-101 fluorescent compound

ABSTRACT: This paper describes a methodology for the exploration of thermal fields on the surface of semiconductor devices, based upon thermal effects on fluorescent films deposited upon the investigated surface. Attention to this method was directed in general by the connection between thermal field patterns and defects in semiconductor devices; and in a more specific way, by the drawbacks of high inertia of other feasible methods, such as e.g. evaporographs. The films used in the described method were dried deposits from ethyl alcohol suspensions, based upon ZnS with added activators. Compound K-9 and FK-101 decrease their brightness upon heating. Compound FKP-03K initially increases its brightness by a temporary flash. The apparatus for the exploration of temperature effects on fluorescence of the compounds consisted of a metal ribbon with the deposited compound on one side irradiated by ultraviolet light

Card 1/2

UDC: 539.293.536

ACC NR: AP6013520

and observed by a photomultiplier thru an ultraviolet-opaque filter. A heat source and a thermocouple riding upon the opposite side of the metal ribbon controlled the compound's temperature. It was found possible, using three compounds as required, to cover the temperature range of 20 - 250°C., and to attain adequate sensitivity - a doubling of luminosity for a 10°C temperature fall. With this method, the distributions of surface temperatures can be adequately evaluated quantitatively for the purposes at hand. Transistor and diode surface temperature patterns during overloads and breakdowns are shown. Characteristic hot spots appear e.g. upon the surface of a diode under conditions of an avalanche breakthrough. Orig. art. has 8 figures.

SUB CODE: 20/ SUBM DATE: 03Mar65/ ORIG REF: 000/ OTH REF: 001

Card 2/2



FEN'OR, S. M.

"Problems Arising in the Complex Conservative Method of Treating Female Sterility," Akusher i Ginekol., No.5, 1949.

Hon. Dr., Turkmen SSR  
Obstetrical and Gynecological Clinic, Turkmen State Inst. Neurology and Physiotherapy.

Fen'or, S.M.  
FEN'OR, S.M.

Gordeev's liquid for treating endocervicitis and erosions of the  
cervix uteri. Sov.med. 21 Supplement:28 '57. (MIRA 11:2)

1. Iz kliniki detskikh bolezney Rostovskogo meditsinskogo instituta.  
(UTERUS--DISEASES)

FEN'OR, S.M., zasluzhennyy vrach Trukmenskoy SSR

Method for treating endocervicitis and erosion of the cervix uteri  
with Gordeev's solution No.2. Zdrav. Turk. 5 no.4:35-38 J1-Ag '61.

(MIRA 14:10)

(UTERUS--DISEASES)

(SOLUTIONS (PHARMACY))

FENOSHINA, U.I.

Glaucinite from Lower Tortonian sediments of the Lyuben' Velikiy  
health resort. Vop. min. osad. obr. 6:266-282 '61. (MIRA 15:6)  
(Lvov region--Glaucinite)

FENOSHINA, U.I.

New data on skolite. Vop. min. csad. obr. 6:283-295 '61.  
(Skolite) (MIRA 15:6)

PAKHUTSKIY, Ch.; LAZARENKO, Ye.K.; FENOSHINA, U.I.

Glaucosite from Cretaceous sediments of Ruzanki (southeastern  
part of the central Sudetes). Vop. min. osad. obr. 6:333-348  
'61. (MIRA 15:6)

(Sudetes--Glaucosite)

FENOSHINA, U.I.; DRITS, V.A.

Structure of skolite. Min. sbor. no.15:255-261 '61. (MIRA 15:6)

1. Gosudarstvennyy universitet imeni Ivana Franko, L'vov i  
Gosudarstvennyy universitet imeni A.A. Zhdanova, Irkutsk.  
(Skolite)

FENCOSHINA, U.I.; PAVLISHIN, V.I.

Feodorov Scientific Session. Min. sbor. no.15:448-449 '61.  
(MIRA 15:6)

1. Gosudarstvennyy universitet, L'vov.  
(Mineralogy)



LANGER-KUZ'NYAROVA, A.; LAZARENKO, Ye.K.; PENOSHINA, U.I.

Mineralogy of the Ordovician glauconitite of Tluszcz. Min. sbor. no.  
17:170-176 '63. (MIRA 17:11)

1. Geologicheskii institut, Varshava i Gosudarstvennyi universitet  
imeni Franko, L'vov.

MACKIEWICZ, Stefan; FENRYCH, Wladyslaw

Immuno-electrophoresis. Postery hig.med.dosw. 13 no.6:809-816  
'59.

(BLOOD PROTEINS chem)  
(ELECTROPHORESIS)

FENRYCH, Wladyslaw; JAZIENICKI, Boguslaw; MACKIEWICZ, Stefan; MACKIEWICZ,  
Urszula; TWARDOWSKI, Krzysztof

Separation of human serum proteins by means of an immuno-chemical  
method. Polski tygod. lek. 14 no. 44:1937-1939 2 Nov 59.

1. (Z III Kliniki Chorob Wewnętrznych A. M. w Poznaniu; kierownik:  
prof. dr F. Labendzinski i z pracowni Farmakodynamiki A. M. w  
Poznaniu; kierownik: prof. dr J. Dąbrowski).  
(BLOOD PROTEINS, chem.)

FENRYCH, Wladyslaw; JAZIENICKI, Boguslaw; ROZWADOWSKA-DOWZENKO, Maria

Observation on the immuno-electrophoretic picture of serum proteins in patients treated with adrenal cortex hormones. Polskie arch. med.wewn. 30 no. 6:792-794 '60.

1. Z III Kliniki Chorob Wewnętrznych A.M. w Poznaniu P.o. Kierownika  
Kliniki: doc. dr med. M. Rozwadowska-Dowzenko  
(ADRENAL CORTEX HORMONES pharmacol)  
(BLOODPROTEINS pharmacol)

MACKIEWICZ, Stefan; FENRYCH, Wladyslaw; JAZIENICKI, Boguslaw; MACKIEWICZ, Aniela

Immune-electrophoretic studies on proteins in the blood serum and exudates in patients with rheumatism. Reumatologia Polska no.3: 129-136 '60.

1. Z III Kliniki Chorob Wewnętrznych AM w Poznaniu Kierownik: prof. dr F. Labandzinski Z Pracowni Farmakodynamiki AM w Poznaniu Kierownik: prof. dr J. Dadles

(RHEUMATISM metabolism)  
(BLOOD PROTEINS)  
(EXUDATES AND TRANSUDATES metab)  
(PROTEINS metab)

FENRYCH, W.; PAWELKIEWICZ, J.; MAGAS, S.

Conversion of cyanocobalamin, in vivo, into its coenzyme form  
in the rabbit. Bul Ac Pol biol 10 no.4:117-119 '62.

1. Department of Biochemistry, College of Agriculture, Poznan  
Department II of Internal Diseases, School of Medicine, Poznan.  
Presented by J.Heller.

\*

FENSTER, G.S. (Leningrad)

Effect of preliminary adaptation to oxygen deficiency on the  
course of burns. Pat. fiziol. i eksp. terap. 4 no. 5:57-58  
S-0 '60. (MIRA 13:12)

1. Iz patofiziologicheskoy ozhogovoy laboratorii (nachal'nik-  
dotsent Ye.V. Gubler) kafedry gosital'noy khirurgii No. 1  
(nachal'nik - prof. I.S. Kolesnikov) Voenno-meditsinskoy  
ordena Lenina akademii imeni S.M. Kirova.  
(BURNS AND SCALDS) (ANOXEMIA)

SKORIK, V.I.; KOCHETYGOV, N.I.; KONSTANTINOV, V.A.; FENSTER, G.S.;  
PENCHUK, V.M. (Leningrad)

Model of burn emaciation in laboratory animals. Pat. fiziol. i  
eksp. terap. 5 no.6:64-65 N-D '61. (MIRA 15:4)

1. Iz Voenno-meditsinskoy ordena Lenina akademii imeni S.M.Kirova.  
(BURNS AND SCALDS)



FENSTER, G.S. (Leningrad)

Effect of contidioning to hypoxia on the course of burn disease.  
Pat fiziol. eksp. ter. 7 no.5:40-43 S-0'63 (MIRA 17:2)

1. Iz nauchno-issledovatel'skoy ozhogovoy lanoratorii ( nachal'-  
nik - doktor med. nauk Ye.V.Gubler) i kafedry termicheskikh  
porazheniy ( nachal'nik - prof. T.Ya Ar'yev) Voenno-meditsein-  
skoy ordena Lenina akademii imeni S.M.Kirova.

KOVAL'CHUK, V.; FENTSIK, I.

Obstacles in the training of miners. Prof.-tekh. obr. 20 no.7:  
29-30 JI '63. (MIRA 16:10)

1. Starshiy inzh. po tekhnicheskoy uchebe rudnika imeni Frunze,  
Krivoy Rog (for Koval'chuk). 2. Starshiy inzh. po tekhnicheskoy  
sluzhbe rudnika imeni Kominterna (for Fentsik).

FEN'VESH, E.; GEMESHI, T.; NEMET, F.; SHANDOR, T.; GASYOROVSKI, L.;  
STARZHINSKI, A.

Semiautomatic measuring instrument for processing pictures obtained  
in the bubble chamber and the Wilson chamber. Prib. i tekh. eksp.  
6 no.2:68-72 Mr-Apr '61. (MIRA 14:9)

1. Tsentral'nyy issledovatel'skiy institut fiziki, Budapesht (for  
Fen'vesh, Gemeshi, Nemet, Shandor). 2. Institut yadernykh  
issledovaniy, Varshava (for Gasyorovski, Starzhinski).  
(Photography, Particle track)

STADNIK, P.M. [Stadnykh, P.M.]; FENTSIK, V.P. [Fentayk, V.P.]

Effect of an electric field on the oxidation of methanol on a silver catalyst. Dop. AN URSR no. 12:1608-1610 '60.

(MIRA 14:1)

1. Uzhgorodskiy gosudarstvennyy universitet. Predstavleno akademikom AN USSR A.M. Brodskim.  
(Oxidation) (Methanol) (Electric fields)

STADNIK, P.M.; FENTSIK, V.P.

Effect of annealing temperature on formaldehyde yield in  
methanol oxidation on a silver catalyst. Zhur. fiz.  
khim. 35 no.7:1425-1429 J1 '61. (MIRA 14:7)

1. Uzhgorodskiy gosudarstvennyy universitet.  
(Formaldehyde) (Methanol)

51190

27208

S/08/61/000/014/003/030  
B106/B410

AUTHORS: Fentsik, V. P., Stadnik, P. M.

TITLE: Effect of an electric field on methanol oxidation with a silver catalyst

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 14, 1961, 74, abstract 145512. (Dokl. i soobshoh. Uzhgorodsk. un-t. Ser. khim., no. 3, 1960, 16-17)

TEXT: The effect of an electric field on  $\text{CH}_3\text{OH}$  oxidation with an Ag catalyst was studied. The voltage was applied between the catalyst and an electrode attached in the middle of the reaction vessel. The electrode temperature was kept near room temperature. If the catalyst was used as cathode, the  $\text{CH}_2\text{O}$  yield rose by 6-7 %, irrespective of the composition of the reaction mixture; if the catalyst was used as anode, the yield dropped by 2-3 %; the yields in  $\text{CO}_2$ , CO, and  $\text{H}_2$  did practically not change. At potential differences of 0-50 v, the  $\text{CH}_2\text{O}$  yield changed

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Effect of an electric field on...

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proportional to the voltage applied; a further increase in the potential difference showed no effect which according to the authors is probably due to saturation of the catalyst surface with free electrons at the expense of their number in the reaction space. [Abstracter's note: Complete translation.]

X

Card 2/2

5.1190

27207  
S/081/61/000/014/002/030  
B106/B110

AUTHORS: Fentsik, V. P., Stadnik, P. M.

TITLE: "Chilling" of the catalytic oxidation of methanol by a solid surface

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 14, 1961, 74, abstract 145511. (Dokl. i soobshch. Uzhgorodsk. un-t. Ser. khim., no. 3, 1960, 18-20)

TEXT: Oxidation of  $\text{CH}_3\text{OH}$  was conducted in a quartz tube lined inside with a silver layer in the central part. The walls of a glass tube attached inside the quartz tube coaxially with the latter were used as chilling surface. The walls of the chilling surface were rinsed with water of constant temperature. The authors studied the effect of the composition of the mixture  $\text{O}_2 + \text{CH}_3\text{OH}$ , the working temperature  $t_1$  of the catalyst, and the temperature  $t_2$  of the chilling surface, on the oxidation rate of  $\text{CH}_3\text{OH}$ . They found that at a certain value of  $t_1$ , and at  $t_2 = 65^\circ\text{C}$

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